# Intercalation of the Niobium-Diselenide Layer Structure by First-Row Transition Metals 

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#### Abstract

The systems $\mathrm{M}_{\mathbf{x}} \mathrm{NbSe}_{2}$ (where $\mathrm{M}=\mathrm{Ti}, \mathrm{V}, \mathrm{Cr}, \mathrm{Mn}, \mathrm{Fe}, \mathrm{Co}, \mathrm{Ni}$ ) were prepared. It appeared that the limit of solid solution was reached at $x \approx \frac{1}{3}$. Crystallographic ordering and changes in the $c$-axis of $\mathrm{NbSe}_{2}$ were observed. Unit cell parameters and crystallographic orderings are compared with those observed in analogous $\mathrm{NbS}_{2}$ systems.


## Introduction

A recent study (1) of the structural behavior of $\mathrm{NbS}_{2}$ and $\mathrm{TaS}_{2}$ revealed that the first-row transition metals $\mathrm{Mn}, \mathrm{Fe}, \mathrm{Co}$, and Ni can be inserted into the octahedral holes between the prismatic sulfur layers, characteristic of $\mathrm{NbS}_{2}$ and $\mathrm{TaS}_{2}$. Superstructures, resulting from an ordering of the inserted metals, were formed. A similar behavior could be expected for the selenides. We therefore started an investigation on combinations of the first-row transition metals with $\mathrm{NbSe}_{2}$.

## Experimental

The samples $\mathrm{M}_{x} \mathrm{NbSe}_{2}\left(x=0 \rightarrow \frac{1}{2}\right)$ were prepared by firing pressed pellets, of the appropriate mixtures of the elements, in sealed evacuated quartz tubes at 800 C for 48 hr . The tubes were allowed to cool in the furnace.

The cell dimensions were calculated from the back reflections of Debye-Scherrer photographs, using a least-squares program. (M. H. Mueller and L. Heaton, report no. ANL-6176). $\mathrm{CuK}_{\alpha 1}$ radiation ( $\lambda=1.54051$ ) was used and the reflections were corrected for shrinkage of the film and absorption, using the Nelson-Riley's extrapolation.

In a number of samples small traces $(\leqslant 5 \%)$ of first-row transition metal selenides could not be avoided, even after regrinding and annealing. It appeared that this does not influence the lattice constants significantly.

## Structural Properties

Diffractometer diagrams $\left(\mathrm{CuK}_{\alpha}\right.$ radiation) of $\mathrm{M}_{x} \mathrm{NbSe}_{2}$ showed that the transition metals $\mathrm{V}, \mathrm{Cr}$,
$\mathrm{Mn}, \mathrm{Fe}, \mathrm{Co}$, and Ni can be inserted between the prismatic $\mathrm{NbSe}_{2}$ layers; only Ti showed much less tendency to enter the lattice. The inclusion of these metals could be concluded from changes in the $c$-axis of $2 s-\mathrm{NbSe}_{2}$ and the observation of weak superlattice reflections, pointing to a possible ordering of the transition-metal atoms. These observations, roughly similar to those in $\mathrm{NbS}_{2}$, gave us reason to assume that the metal atoms enter the octahedra between the $\mathrm{NbSe}_{2}$ layers (Fig. 1). The stronger reflections could be indexed on the original hexagonal lattice of the $2 s-\mathrm{NbSe}_{2}$ type (2). At concentrations of $x$ varying from $\frac{1}{4}$ to $\frac{1}{3}$ weak superlattice reflections showed up corresponding to a unit cell with an axis $2 a$ and not with an axis $a \sqrt{3}$ as observed in the sulfides for $x=\frac{1}{3}$. The periodicity in the $c$-direction was the same as in $2 \mathrm{~s}-\mathrm{NbSc}_{2}$. From the possible orderings it was found that with this $2 a$-supercell a well-defined ordered compound could be expected at $x=\frac{1}{4}$, as shown in Fig. 2. At this concentration one quarter of the octahedra are occupied. For the $2 a$-superlattice, the systematic extinctions $h h l, l=2 n+1$ were observed. Using the space group $\mathrm{P}_{3} / \mathrm{mmc}$ of $2 s$ - $\mathrm{NbSe}_{2}$, the ordered structure for $\mathrm{M}_{1 / 4} \mathrm{NbSe}_{2}$ can be described on the following initial positions:

$$
\begin{aligned}
& 12 \mathrm{~S} \text { in } x, 2 x, z(12 k) \text { with } x=\frac{1}{6}, z=\frac{1}{8} . \\
& 4 \mathrm{~S} \text { in } \frac{1}{3}, \frac{2}{3}, z(4 f) \text { with } z=\frac{5}{8} \\
& 6 \mathrm{Nb} \text { in } \frac{3}{2}, 0,0(6 g) \text {. } \\
& 2 \mathrm{Nb} \text { in } 0,0,0(2 a) \text {. } \\
& 2 \mathrm{M} \text { in } 0,0, \frac{1}{4}(2 b) \text {. }
\end{aligned}
$$

With M in the position ( $2 b$ ) the super-reflections $h k l(l=2 n+1)$ would be extinguished. It proved


Fic. 1. Stacking of prismatic layers in $2 s-\mathrm{NbSe}_{2}$.

- $\left.M\left(1 / 2^{2 / 3} 3^{1 / 4},{ }^{2 / 3 / 3}\right)^{3 / 4}\right)$
- $\mathrm{M}\left(001 / 4,00^{3 / 4}\right)$
- Vacant octahedral holes
-.-.- supercell a $\sqrt{3}$
----- SUPERCELL 20
- original cell


## PRISMATIC Se



Fig. 2. Possible orderings of inserted metal atoms $(z=\downarrow)$ between two prismatic layers.
more difficult to get all the samples completely ordered at $x=\frac{1}{4}$, and up till now annealing at various temperatures did not change the superlattice reflections significantly. Apparently, a higher concentration made it easier to completely fill one crystallographic position, the rest being randomly distributed. We observed the same phenomenon at $x=\frac{1}{2}$, where in the systems containing $\mathrm{Mn}, \mathrm{Fe}, \mathrm{Co}$, and Ni superlattice reflections corresponding to an axis $a \sqrt{3}$ were observed. This type of ordering corresponds to the concentration $x=\frac{1}{3}$, and is extensively discussed in a previous paper (1). This behavior shows that possibly a concentration higher than $x=\frac{1}{3}$ can be inserted in the four mentioned systems. An example of this is shown in Table III where the low-angle reflections for $\mathrm{Fe}_{1 / 3} \mathrm{NbSe}_{2}$ and $\mathrm{Fe}_{1 / 2} \mathrm{NbSe}_{2}$ are listed. Most of the superlattice reflections were observed below $\theta=27$.

TABLE I
Latitice Constants of the Systems $\mathrm{M}_{\boldsymbol{x}} \mathrm{NbSe}_{\mathbf{2}}$

|  | $a$ |  |
| :--- | :--- | :--- |
|  |  |  |
|  |  |  |
| $\mathrm{~V}_{1 / 4} \mathrm{NbSe}_{2}$ | 3.451 | 12.660 |
| $\mathrm{~V}_{1 / 3} \mathrm{NbSe}_{2}$ | 3.454 | 12.673 |
| $\mathrm{Cr}_{1 / 4} \mathrm{NbSe}_{2}$ | 3.452 | 12.570 |
| $\mathrm{Cr}_{1 / 3} \mathrm{NbSe}_{2}$ | 3.452 | 12.567 |
| $\mathrm{Mn}_{1 / 4} \mathrm{NbS}_{2}$ | 3.462 | 12.899 |
| $\mathrm{Mn}_{1 / 3} \mathrm{NbSe}_{2}$ | 3.469 | 13.033 |
| $\mathrm{Mn}_{1 / 2} \mathrm{NbSe}_{2}$ | 3.476 | 13.165 |
| $\mathrm{Fe}_{1 / 4} \mathrm{NbSe}_{2}$ | 3.454 | 12.623 |
| $\mathrm{Fe}_{1 / 3} \mathrm{NbSe}_{2}$ | 3.459 | 12.671 |
| $\mathrm{Fr}_{1 / 2} \mathrm{NbSe}_{2}$ | 3.468 | 12.773 |
| $\mathrm{Co}_{1 / 4} \mathrm{NbSe}_{2}$ | 3.452 | 12.379 |
| $\mathrm{Co}_{1 / 3} \mathrm{NbSe}_{2}$ | 3.456 | 12.384 |
| $\mathrm{Ni}_{1 / 4} \mathrm{NbSe}_{2}$ | 3.453 | 12.419 |
| $\mathrm{Ni}_{1 / 3} \mathrm{NbSe}_{2}$ | 3.456 | 12.413 |
| $\mathrm{NbSe}_{2}$ | 3.442 | 12.54 |

## TABLE II

Comparison of c-axes of $\mathrm{M}_{1 / 3} \mathrm{NbS}_{2}$ and $\mathrm{M}_{1 / 3} \mathrm{NbSe}_{2}$;

$$
\Delta=c_{(x=1 / 3)}-c_{(x=0)}
$$

| M | $\mathrm{NbSe}_{2}$ |  | $\mathrm{NbS}_{2}(1)$ |  |
| :---: | :---: | :---: | :---: | :---: |
|  | c | $\Delta$ | c | $\Delta$ |
| None | 12.54 |  | 11.89 |  |
| Mn | 13.033 | 0.493 | 12.629 | 0.739 |
| Fe | 12.671 | 0.131 | 12.212 | 0.322 |
| Co | 12.384 | -0.156 | 11.886 | -0.004 |
| Ni | 12.413 | -0.137 | 11.897 | 0.007 |

TABLE III
Low-Angle Reflections from Diffractograms ( $\mathrm{CuK}_{\alpha} ; \theta=10-27^{\circ}$ )

| $\mathrm{Fe}_{1 / 3} \mathrm{NbSe}_{2}$ <br> Systematic extinctions ( $h h l$ ), $l=2 n+1$. <br> $\mathrm{P} 63 / \mathrm{mmc}$, indexed on $2 a$-supercell |  |  |  | $\begin{gathered} \mathrm{Fe}_{1 / 2} \mathrm{NbSe}_{2} \\ \text { Systematic extinctions }(00 l), l=2 n+1 . \\ \text { P63 }_{3} 22 \text {; indexed on } a \sqrt{3} \text {-supercell } \end{gathered}$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $10^{4} \sin ^{2} \theta_{\text {obs }}$ | $10^{4} \sin ^{2} \theta_{\text {catc }}$ | $h k l$ | $J_{\text {obs }}$ | $10^{4} \sin ^{2} \theta_{\text {obs }}$ | $10^{4} \sin ^{2} \theta_{\text {calc }}$ | $h k l$ | $I_{\text {abs }}$ |
| 311 | 314 | 102 | w | 365 | 365 | 102 | vw |
| 496 | 497 | 1103 | vw | 548 | 547 | 103 | w |
|  | 499\} | $103\}$ |  | 583 | 583 | 004 | w |
| 592 | 592 | 004 | w | 659 | 659 | 110 | mw |
| 645 | 645 | 112 | vw | 695 | 695 | 111 | st |
| 663 | 663 | 200 | w |  | 802 | 104 |  |
|  |  |  |  | 805 |  |  | st |
| 698 | 699 | 201 | st |  | 804 | 112 |  |
| 756 | 758 | 104 | w | - | 878 | 200 | - |
| 809 | 811 | 202 | st | 916 | 915 | 201 | w |
| 994 | 996 | 203 | st | 987 | 986 | 113 | st |
|  | 1089 | 114 |  | 1024 | 1024 | 202 | vw |
| 1089 |  |  | w |  |  |  |  |
|  | 1091 J | 105 |  | 1130 | 1130 | 105 | w |
| 1161 | 1159 | 210 | vw | - | 1206 | 203 | - |
| 1252 | 1255 | 204 | vst | 1242 | 1241 | 114 | vst |
| 1300 | 1307 | 212 | vw | 1312 | 1311 | 006 | mw |
| 1331 | 1333 | $\begin{aligned} & 006 \\ & 300 \end{aligned}$ | m | - | 1461 | 204 | - |
|  | 1491 |  |  |  | 1531 | 106 |  |
| - | 1492 ] | 213 | - . | - | 1537 ( | 210 | - |
| 1499 | 1498 | 106 | w |  | 1570 | 115 |  |
| - |  |  |  | 1570 |  |  | m |
| - | 1527 | 301 | - |  | 1574 ( | 211 J |  |
| 1588 | 1588 | 205 | mst | - | 1683 | 212 | - |
| 1640 | 1639 | 302 | vw | - | 1789 | 205 | - |
| - | 1751 | 214303 | - | 1866 | 1865 | $\begin{aligned} & 213 \\ & 116 \end{aligned}$ | vw |
|  | 1824 ) |  |  |  | 1970 |  |  |
| 1818 | 1829 | 116 | w | 1977 | 1976 | 300 | st |
| - | 1979 | 107 | - | 609 |  |  | w |
|  | 1987 | 220 |  |  |  | FeSe |  |
| 1988 | $1995\}$ | 206 | st |  |  |  |  |
| 775 |  | ? | w |  |  |  |  |

The intensities of $\mathrm{Fe}_{1 / 2} \mathrm{NbSe}_{2}$ show roughly the same trend as those of $\mathrm{Fe}_{1 / 3} \mathrm{NbS}_{2}$ (1). A largely similar pattern was observed for the rest of the systems.

In order to determine the influence of the inclusion of metal atoms on the $\mathrm{NbSe}_{2}$ lattice, the cell constants of the subcell were calculated. For those systems, where the concentration $x=\frac{1}{2}$ did not show any change in axes compared to the lower concentrations, these calculations were omitted.

From the results listed in Table I it appears that in the Mn and Fe system the concentration is indeed
higher than $x=\frac{1}{3}$, as in these cases there is an expansion of the lattice up to $x=\frac{1}{2}$. As the sample $\mathrm{Fe}_{1 / 2} \mathrm{NbSe}_{2}$ contains an excess of ferromagnetic iron, the maximum concentration with our experimental conditions must be somewhere between $x=\frac{1}{3}$ and $\frac{1}{2}$. As a check, a number of samples $\mathrm{Fe}_{x} \mathrm{NbSe}_{2}$ ( $x>\frac{1}{2}$ ) were prepared. The lattice constants appeared to be the same as for $\mathrm{Fe}_{1 / 2} \mathrm{NbSe}_{2}$ and an increase of ferromagnetic iron as function of concentration was observed. From the fact that the samples $\mathrm{V}_{1 / 2} \mathrm{NbSe}_{2}$ and $\mathrm{Cr}_{1 / 2} \mathrm{NbSe}_{2}$ contained a considerable amount of V and Cr selenides, a maximum concentration of
$x \leqslant \frac{1}{3}$ may be concluded. This is also evident from the much slower increase of the $c$-axis after an initial expansion at $x=\frac{1}{4}$.

The lattice constants of the compounds $\mathrm{M}_{1 / 3} \mathrm{NbSe}_{2}(\mathrm{M}=\mathrm{Mn}, \mathrm{Fe}, \mathrm{Co}, \mathrm{Ni})$ are compared with those of the corresponding sulfides (Table II). In $\mathrm{NbS}_{2}, \mathrm{Mn}$, and Fe cause an expansion of the lattice in the $c$-direction. Ni causes a small expansion and Co causes a slight contraction of the lattice. In the selenides (as compared to the sulfides) Mn and Fe bring about smaller expansions of the lattice. For Co , a larger contraction of the lattice is observed and for Ni a change from a small expansion in the sulfide to a contraction of the lattice in the selenide. These differences are undoubtedly caused by the presence of larger octahedral holes in $\mathrm{NbSe}_{2}$ than in $\mathrm{NbS}_{2}$. The effective paramagnetic moment of Mn in $\mathrm{Mn}_{1 / 3} \mathrm{NbSe}_{2}$ as obtained from magnetic measurements (inverse susceptibility vs. temperature) is $4.90 \mu_{\mathcal{B}}$. This is in good agreement with the spinonly moment $4.91 \mu_{B}$, calculated for $\mathrm{Mn}^{3+}$. Roomtemperature Mössbauer spectra indicate the iron in $\mathrm{Fe}_{1 / 3} \mathrm{NbSe}_{2}$ is $\mathrm{Fe}^{2+}$. Crystal growth of these materials is being carried out by vapour transport
methods in order to further study their crystallographic and magnetic properties.

## Summary

Combinations of $\mathrm{NbSe}_{2}$ with first-row transition metals were prepared. Of the latter the metals $\mathrm{V}, \mathrm{Cr}$, $\mathrm{Mn}, \mathrm{Fe}, \mathrm{Co}$, and Ni enter the octahedral holes between the prismatic layers of $\mathrm{NbSe}_{2}$. Two types of ordered distributions ( $2 a$ - and $a \sqrt{3}$-supercells of $2 s-\mathrm{NbSe}_{2}$ ) were observed. The influence of inclusion of the transition metals on the lattice was consistent with the difference in anion sizes of $\mathrm{NbS}_{2}$ and $\mathrm{NbSe}_{2}$.

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## References

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